## The Stereochemistry of a Cyclic Sulphite of $10\beta$ -Pinane-2,3 $\alpha$ -diol

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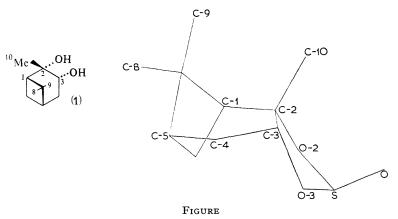
REACTION<sup>1</sup> of  $(\pm)$ -10 $\beta$ -pinane-2,3 $\alpha$ -diol (1) with thionyl chloride gave a mixture of two cyclic sulphites, separable by fractional crystallization. Both isomers (A, m.p. 50—52°; B, m.p. 102—103°) gave the diol (1) on mild basic hydrolysis. Two isomers could arise if the 10-Me group were *cis*- or *trans*- to the extracyclic sulphite oxygen atom. A preliminary X-ray crystal-structure analysis has established the relative configuration of isomer (A) and the solid state conformation of a *vicinal*-diol cyclic sulphite ring-system shown in the Figure.

Crystals of the cyclic sulphite (A) grown from pentane are triclinic, space group  $P\bar{1}$ , a = 7.42, b = 10.94, c = 7.18 Å,  $\alpha = 109.1^{\circ}$ ,  $\beta = 95.2^{\circ}$  and  $\gamma = 90.2^{\circ}$ , Z = 2. They are often twinned and sublime readily at 20°. Three-dimensional intensity data were recorded at 4° with a Nonius integrating Weissenberg camera and a crystal sealed in a capillary. The positions of all non-hydrogen atoms were established by conventional Patterson, difference Fourier, and least-squares techniques using 680 reflections, the intensities of which were measured photometrically above background; R = 18.

The n.m.r. data for isomer (A) are consistent with the existence of the cyclic sulphite in essentially the same

\* Determined at 60 Mc./sec. for 20% solutions (w/v) in CDCl<sub>3</sub> with CHCl<sub>3</sub> and Me<sub>4</sub>Si as internal standards.

conformation in both the solid and liquid phase. The notable feature of the n.m.r. data for the cyclic sulphites A and B, epimeric at the sulphur atom, is the markedly downfield location of the signals due to the 10-Me and 3-H functions in isomer (A) relative to isomer (B). The deshielding of the 10-Me and 3-H in isomer (A) by the  $S \rightarrow O$  group



is analogous to the "syn-axial effect" encountered<sup>2</sup> in the n.m.r. study of six-membered-ring sulphoxides. Clearly, however, reservations about the application of this concept

alone to the interpretation<sup>3</sup> of the n.m.r. spectra of sulphoxides must apply also in this study.

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